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Work The diagram will then show you that the isotopes and ions are in your soil. You will then need to write down the isotopes and ions that you have listed.

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Answer: 126.86 amu 10. The natural abundance for boron isotopes is 19.9% ^{10}B and 80.1% ^{11}B . Calculate boron's atomic mass. Answer: 126.86 amu 11. Hydrogen is 99% ^1H , 0.8% ^2H , and 0.2% ^3H . Calculate its average atomic mass. Answer: 1.21 amu 12. Rubidium is a soft,

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silvery-white metal that has two common isotopes, ^{85}Rb and ^{87}Rb . If the ...

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Answer Key

For each of the following ions, indicate the total number of protons and electrons in the ion.

Ion	Number of Protons	Number of Electrons
Co^{+2}	27	25
Co^{+3}	27	24
Cl^{-1}	17	18
K^{+1}	19	18
S^{-2}	16	18
Sr^{+2}	38	36
Al^{+3}	13	10
P^{-3}	15	18

How many protons, neutrons, and electrons are present in the $^{59}\text{Ni}^{+2}$ ion?

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P: 28 N: 31 E: 26 28

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WORKSHEET Atomic # = # of protons.

Mass # = Atomic # +

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neutrons form the nucleus or central region of the atom. Since neutrons do not influence the charge, the range of neutrons isn't based on the variety of protons and will vary even among atoms of the same element.

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